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MASS SPECTROMETRIC STUDY OF MODIFIED URIDINES AND THEIR N3-ISOMERS

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Abstract: The mass spectral fragmentations of modified uridines and their N^3 -isomers are discussed in context of the b+41 ion formation.

In the present work the electron impact mass spectra of modified uridines (1) and their N3-isomers (2) (Figure 1) have been analyzed.

The main fragmentation pathways of all investigated compounds correspond to known EI fragmentation of underivatized nucleosides. However, in the mass spectra of N3-isomers (2) an interesting peak b+41 (at m/z 41 units higher than the peak corresponding to the pyrimidine base) has been observed. The intensity of the b+41 ion may be used to

FIGURE 1. Structures of modified uridines.

TABLE	1.	Relative	intensity	of	the	b+41	ion	in	the	mass	
		spectra o	of modified	l ur	idir	nes a	nd tl	hein	c N3-	-isome	rs

Co	mpoun	 d	70	e V	15 eV		
Rı	Rz	X	N1-isomer	N3-isomer	N1-isomer	N3-isomer	
Н	Н	0	1.8	6.4	1.6	11.1	
СНз	Н	0	1.0	8.3	0.4	14.0	
ОСНз	Н	0	0.2	1.4	0.1	1.6	
Н	СНз	0	3.5	0.8	4.1	9.0	
Н	H	S	26.9	11.5	27.2	16.3	
ОСНз	Н	S	4.6	1.5	4.8	1 . 8	

differentiate N^{1} - from N^{3} -isomers of modified uridines (Table 1).

The detection of the b+41 ion in the mass spectra of N³-isomers of modified uridines and in model compounds: N- β -D-ribofuranosylpyrimidone-2, N- β -D-ribofuranosylpyridone-2 as well as in 2-thiouridines¹ provides some additional information concerning the mechanism which has been proposed for the production of this ion in the fragmentation of cytidines² and 3-deazauridine³. The carbonyl or thiocarbonyl group adjacent to the N-glycosidic bond and the conjugated double bond seem to be necessary for the b+41 ion formation. The sulphur atom at the 2-position significantly facilitates production of this ion, while chemical nature of substituents at 4-, 5- or 6-position has relatively less influence.

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